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APPLICATION NO.	FILING DATE		FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
09/463,423 04/07/2000		/07/2000	GING HAUW KHOE	GRIHACP28AU	6300 21
20210	7590	11/19/2003 EXAMINER			
DAVIS & E FOURTH FL	,	P.L.L.C.	CHORBAJI, MONZER R		
500 N. COM		STREET	ART UNIT	PAPER NUMBER	
MANCHEST	ER, NH	03101-1151	1744		

DATE MAILED: 11/19/2003

Please find below and/or attached an Office communication concerning this application or proceeding.

					nv				
•		Application	n No.	Applicant(s)					
	_	09/463,423	3	KHOE ET AL.					
	Office Action Summary	Examiner		Art Unit	<u>-</u> .				
		MONZER F	R CHORBAJI	1744					
Period fo	The MAILING DATE of this communication reply	n appears on the	cover sheet with the	correspondence add	iress				
THE - Exte after - If the - If NO - Failu - Any	ORTENED STATUTORY PERIOD FOR R MAILING DATE OF THIS COMMUNICATI nsions of time may be available under the provisions of 37 C SIX (6) MONTHS from the mailing date of this communicative period for reply specified above is less than thirty (30) days period for reply is specified above, the maximum statutory period for reply within the set or extended period for reply will, by reply received by the Office later than three months after the ed patent term adjustment. See 37 CFR 1.704(b).	ON. CFR 1.136(a). In no ever on. to a reply within the statuly period will apply and will statute, cause the applic	nt, however, may a reply be to cory minimum of thirty (30) da expire SIX (6) MONTHS from cation to become ABANDON	imely filed ays will be considered timely in the mailing date of this con ED (35 U.S.C. § 133).	mmunication.				
1)🖂	Responsive to communication(s) filed on	06 August 2003.							
2a) <u></u>	This action is FINAL . 2b)⊠	This action is not	n-final.						
3)	Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under Ex parte Quayle, 1935 C.D. 11, 453 O.G. 213.								
Disposit	ion of Claims								
4)⊠	☑ Claim(s) <u>10-29</u> is/are pending in the application.								
	4a) Of the above claim(s) is/are withdrawn from consideration.								
5)□	Claim(s) is/are allowed.								
6)⊠	Claim(s) <u>10-29</u> is/are rejected.								
7)	Claim(s) is/are objected to.								
8)[Claim(s) are subject to restriction a	and/or election re	quirement.						
Applicat	ion Papers								
9)[The specification is objected to by the Exa	aminer.							
10)🏻	10)⊠ The drawing(s) filed on <u>07 March 2000</u> is/are: a)⊠ accepted or b)□ objected to by the Examiner.								
	Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).								
_	Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).								
11)[The oath or declaration is objected to by the	he Examiner. Not	e the attached Offic	e Action or form PT	O-152.				
Priority (under 35 U.S.C. §§ 119 and 120								
* (13)	Acknowledgment is made of a claim for for All b) Some * c) None of: 1. Certified copies of the priority docu 2. Certified copies of the priority docu 3. Copies of the certified copies of the application from the International Besee the attached detailed Office action for Acknowledgment is made of a claim for doince a specific reference was included in the Terminal Communication of the foreign language Acknowledgment is made of a claim for doince a specific reference was included in the Communication of the foreign language Acknowledgment is made of a claim for doing the second secon	ments have been ments have been priority docume Bureau (PCT Rule a list of the certification mestic priority un he first sentence ge provisional appressic priority un	n received. In received in Applicants have been received 17.2(a)). It is decopies not received a 15.C. § 119 of the specification of the specification for the 15 U.S.C. §§ 12	tion No ved in this National S ved. (e) (to a provisional or in an Application I eceived. 0 and/or 121 since a	application) Data Sheet.				
r	eference was included in the first sentence	e of the specificat	on or in an Applicati	ion Data Sheet. 37 (CFR 1.78.				
Attachmen									
2) Notice	ce of References Cited (PTO-892) ce of Draftsperson's Patent Drawing Review (PTO-94 mation Disclosure Statement(s) (PTO-1449) Paper N	18)	4) Interview Summar 5) Notice of Informal 6) Other:						

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DETAILED ACTION

This non-final office action is in response to the amendment received on 08/06/2003

Claim Rejections - 35 USC § 112

1. The following is a quotation of the first paragraph of 35 U.S.C. 112:

The specification shall contain a written description of the invention, and of the manner and process of making and using it, in such full, clear, concise, and exact terms as to enable any person skilled in the art to which it pertains, or with which it is most nearly connected, to make and use the same and shall set forth the best mode contemplated by the inventor of carrying out his invention.

2. Claim 26 is rejected under 35 U.S.C. 112, first paragraph, as failing to comply with the written description requirement. The claim(s) contains subject matter, which was not described in the specification in such a way as to reasonably convey to one skilled in the relevant art that the inventor(s), at the time the application was filed, had possession of the claimed invention.

In claim 26, line 6; applicant uses the phrase "without use of any chelate". The original disclosure does not include such a limitation.

Claim Rejections - 35 USC § 102

3. The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless -

- (b) The invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States.
- 4. Claims 10-29 are rejected under 35 U.S.C. 102(b) as being anticipated by Khoe (WO 95/11195).

With respect to claims 10, 18, and 26, Khoe discloses a method for oxidizing various types of organic species (abstract, lines 1-2 and page 6, lines 7-27) such as

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arsenic and iron. The method of Khoe is not only limited to oxidizing arsenic or iron, the method is expected to oxidize any type of inorganic species present in an aqueous solution. The method includes the following steps: supplying an oxidizable source of sulfur (page 6, line 22) and oxygen (page 6, lines 31-32) to the solution and irradiating the solution with UV light (page 6, lines 18-19) in order to reduce a toxicity of the inorganic species (abstract, lines 3-4). In addition, Khoe forms an aqueous solution, which includes industrial waste water (page 3, lines 18-26).

With respect to claims 11, 19, and 27, Khoe teaches that since the source of sulfur is in an aqueous solution, then it is inherent that sulfur will be in various ionic forms (page 6, lines 10-11).

With respect to claims 12-13, 20-21, and 28, Khoe teaches that inorganic species such as arsenic or iron are in trace quantities (page 16, lines 2-5) of drinking water such that the method of Khoe is not only limited to oxidizing arsenic or iron, the method is expected to oxidize any type of inorganic species present in an aqueous.

With respect to claims 14-16, 22-24, and 29, Khoe discloses the following: the wavelength of UV light is less than 300nm (page 8, lines 11-12), dissolved oxygen is derived from air (page 20, lines 28-30) and the dissolved oxygen is derived from a gas source with an oxygen partial pressure of about 0.2 atmospheres (page 21, lines 1-4).

With regard to claims 17 and 25, Khoe discloses the aqueous solution is industrial waste water (page 3, lines 18-26).

Claim Rejections - 35 USC § 103

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5. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

- 6. The factual inquiries set forth in *Graham* v. *John Deere Co.*, 383 U.S. 1, 148 USPQ 459 (1966), that are applied for establishing a background for determining obviousness under 35 U.S.C. 103(a) are summarized as follows:
 - 1. Determining the scope and contents of the prior art.
 - 2. Ascertaining the differences between the prior art and the claims at issue.
 - 3. Resolving the level of ordinary skill in the pertinent art.
 - 4. Considering objective evidence present in the application indicating obviousness or nonobviousness.
- 7. This application currently names joint inventors. In considering patentability of the claims under 35 U.S.C. 103(a), the examiner presumes that the subject matter of the various claims was commonly owned at the time any inventions covered therein were made absent any evidence to the contrary. Applicant is advised of the obligation under 37 CFR 1.56 to point out the inventor and invention dates of each claim that was not commonly owned at the time a later invention was made in order for the examiner to consider the applicability of 35 U.S.C. 103(c) and potential 35 U.S.C. 102(e), (f) or (g) prior art under 35 U.S.C. 103(a).
- 8. Claims 10-29 are rejected under 35 U.S.C. 103(a) as being unpatentable over Khoe (WO 95/11195).

With respect to claims 10, 18, and 26, Khoe discloses a method for oxidizing various types of inorganic species (abstract, lines 1-2 and page 6, lines 7-27) such as

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arsenic and iron. The method of Khoe is not only limited to oxidizing arsenic or iron, the method is intrinsically capable of oxidizing any type of inorganic species present in an aqueous solution such that iron, arsenic manganese, nickel or any inorganic species can be oxidized by the Khoe method. The method includes the following steps: supplying an oxidizable source of sulfur (page 6, line 22) and oxygen (page 6, lines 31-32) to the solution and irradiating the solution with UV light (page 6, lines 18-19) in order to reduce a toxicity of the inorganic species (abstract, lines 3-4). In addition, Khoe forms an aqueous solution, which includes industrial waste water (page 3, lines 18-26).

With respect to claims 11, 19, and 27, Khoe teaches that since the source of sulfur is in an aqueous solution, then it is intrinsic that sulfur will be in various ionic forms (page 6, lines 10-11).

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With respect to claims 14-16, 22-24, and 29, Khoe discloses the following: the wavelength of UV light is less than 300nm (page 8, lines 11-12), dissolved oxygen is derived from air (page 20, lines 28-30) and the dissolved oxygen is derived from a gas source with an oxygen partial pressure of about 0.2 atmospheres (page 21, lines 1-4).

With regard to claims 17 and 25, Khoe discloses the aqueous solution is industrial waste water (page 3, lines 18-26).

Response to Arguments

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9. Applicant's arguments filed 08/06/2003 have been fully considered but they are not persuasive.

On page 4 of the response, applicant argues, "WO 95/11195 does not in any way teach, suggest or disclose sulphur as an oxidizing source or photoabsorber". The WO 95/11195 teaches on page 6, lines 1013 and line 22 that the substance, which is Sulphur, is capable of being oxidized. That means that sulphur is an oxidizable source as disclosed in claims 10, 18, and 26.

On page 4 of the response, applicant argues, "In fact, WO 95/11195 very clearly teaches, suggests and discloses that the photoabsorber is a dissolved cationic metal species such as Fe(II) or Fe(III), Cu2, etc". The limitation "photoabsorber" is not recited in the claims.

On page 4 of the response, applicant argues, "Further, Frame et al. '581 relates to oxidative removal of cyanide from aqueous streams abetted by UV radiation, however, this oxidation with UV irradiation occurs only when certain metal chelates are used as catalysts". The Frame et al reference was only applied to show that cyanide is a known pollutant that needs to be removed from aqueous solutions. In addition, the Frame et al reference teaches known methods of removing cyanide where metal chelates were not used. Furthermore, the limitation "without use of any chelate" is a new matter since the original disclosure does not recite such a limitation.

Conclusion

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10. Any inquiry concerning this communication or earlier communications from the

examiner should be directed to MONZER R CHORBAJI whose telephone number is

(703) 305-3605. The examiner can normally be reached on M-F 8:30-5:00.

11. If attempts to reach the examiner by telephone are unsuccessful, the examiner's

supervisor, ROBERT J WARDEN can be reached on (703) 308-2920. The fax phone

numbers for the organization where this application or proceeding is assigned are (703)

305-3599 for regular communications and (703) 305-7719 for After Final

communications.

12. Any inquiry of a general nature or relating to the status of this application or

proceeding should be directed to the receptionist whose telephone number is (703) 308-

0661.

Monzer R. Chorbaji MRC

Patent Examiner

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November 17, 2003

Robert J. WARDEN, SR.

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